ABIETANE DITERPENOIDS FROM THE ROOT OF SALVIA PHLOMOIDES

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Key Word Index—Salvia phlomoides; Labiatae; diterpenoids; new abietane derivatives; demethylcryptojaponol; 14-deoxycoleon U; salviphlomone.

Abstract—From the root of *Salvia phlomoides* three new abietane diterpenoids, demethylcryptojaponol, 14-deoxycoleon U and salviphlomone, have been isolated, besides the previously known diterpenes 8,13-abietadiene, 8,11,13-abietatriene, royleanone, 7α -acetoxyroyleanone, taxodione, taxodone, cryptojaponol and sugiol. The structures of demethylcryptojaponol (11,12-dihydroxy-8,11,13-abietatrien-7-one), 14-deoxycoleon U (6,11,12-trihydroxy-5,8,11,13-abietatetraen-7-one) and salviphlomone (6α , 7β -dihydroxy-8,13-abietadiene-11,12-dione) were established by chemical and spectroscopic means.

INTRODUCTION

In a continuation of our studies on the diterpenoid compounds from Salvia spp. [1, 2], we have now investigated the root of S. phlomoides, a species from the aerial part of which several new triterpenoids have been previously isolated [3, 4]. From the root of this plant 11 diterpenic compounds have been isolated, eight of which are the previously known 8,13-abietadiene [5, 6], 8,11,13-abietatriene [7–9], royleanone (12-hydroxy-8,12-abietadiene-11,14-dione) [10, 11], 7α -acetoxyroyleanone (7α -acetoxy-12-hydroxy-8,12-abietadiene-11, 14-dione) [10], taxodione [11-hydroxy-7,9(11),13-abietatriene-6,12-dione] [11], taxodone [6 α ,11-dihydroxy-7,9(11),13-abietatrien-12-one] [11], cryptojaponol (2, 11-hydroxy-12-

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methoxy-8,11,13-abietatrien-7-one) [12–14] and sugiol (12-hydroxy-8,11,13-abietatrien-7-one) [11, 13–15], and the other three are new substances, whose structures are established as 11,12-dihydroxy-8,11,13-abietatrien-7-one (1, demethylcryptojaponol), 6,11,12-trihydroxy-5,8,11,13-abietatetraen-7-one (3, 14-deoxy-coleon U) and 6α ,7 β -dihydroxy-8,13-abietadiene-11,12-dione (6, salviphlomone).

RESULTS AND DISCUSSION

The first of the new diterpenoids (demethylcryptojaponol, 1), $C_{20}H_{28}O_3$, had an IR spectrum which showed phenolic (3590, 3380 broad, cm⁻¹) and aryl-ketone (3020, 3010, 1655, 1635, 1570 cm⁻¹) absorptions. The presence of an o-diphenol moiety conjugated with a ketone function in compound 1 was revealed by its UV spectra obtained after addition of base, aluminium chloride and boric acid (Table 1), which showed characteristic band shifts of this chromophore [16].

However, it was the ¹H NMR spectrum of demethylcryptojaponol that provided the most information and established an abietane structure such as 1 for this new diterpenoid. Effectively, this spectrum (Table 2) showed signals of an isopropyl group attached to an aromatic ring, and of three methyl groups attached to fully substituted sp³ carbon atoms. A one-proton doublet of triplets signal at δ 3.81 was assigned to the C-1 β equatorial proton $(J_{1\beta, 1\alpha} = 12 \text{ Hz}, J_{1\beta, 2}\alpha = J_{1\beta, 2\beta} = 3 \text{ Hz})$, which appears at lower field because of the existence of an oxygenated function on C-11 [17, 18]. The C-14 aromatic proton showed a singlet at δ 8.08, strongly deshielded by the C-7 ketone grouping, as has been shown in cryptojaponol (2) and sugiol [11–15]. Finally, an ABX system at δ 1.85 (1H, dd, $J_{5\alpha,6\beta}$ = 11 Hz, $J_{5\alpha,6\alpha}$ = 6 Hz), 2.67 (1H, dd, $J_{6\alpha,6\beta}$ = 15 Hz) and 2.70 (1H, dd) was assigned to the C-5 methyne and the C-6 methylene groups. Final proof that demethylcryptojaponol has the structure and absolute configuration depicted in formula 1 was obtained by

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+ AlCl₃-HCl + NaOAc-H₃BO₃ **EtOH** + NaOMe + AlCl₃ (4.10)(4.29)218 (4.22)217 (4.22)217 (4.22)218.5 252 (4.16)(4.15)(3.96)259 (4.00)257 (4.18)235 290 (4.02)315 (3.88)(4.21)280 sh (3.64)(3.90)364 (3.94)(3.36)318 sh (3.85)362 sh (3.23) 349 357 (4.02)420 sh (3.65)

(4.22)

(4.12)

(4.20)

(4.00)

(3.70)

(4.09)

221

256

314

409

+

(4.19)

(4.04)

(4.00)

(4.08)

219

266

296

390

†

340 sh

(4.18)

(4.22)

(3.85)

(3.70)

(3.94)

Table 1. UV spectra of compounds 1, 3 and 6 $[\lambda_{max}$ nm $(\log \varepsilon)]$

(4.10)

(3.99)

(3.91)

(3.95)

(3.99)

(3.52)

(3.48)

(1.90)

221

270

295

412

225

257

364

(4.21)

(4.16)

(3.76)

(4.09)

(4.18)

(4.02)

(4.12)

224

270

280

329

380

464

+

216.5

235

290

215

288

343

217

422

582

256 sh

251.5

3

364 sh

Table 2. ¹H NMR data of compounds 1, 3 and 6 (90 MHz, pyridine-d₅, TMS as internal standard)*

	1	3	6
Η-1β	3.81 dt	≅ 3.50 §	2.76 dı
	$J_{1\beta,1\alpha} = 12 \text{ Hz}$		$J_{1\beta, 1z} = 12.6 \text{ Hz}$
	$J_{1\beta, 2\alpha} = J_{1\beta, 2\beta} = 3 \text{ Hz}$		$J_{1\beta, 2\alpha} = J_{1\beta, 2\beta} = 3 \text{ Hz}$
H-5	1.85 dd		1.53 d
	$J_{5\alpha, 6\beta} = 11 \text{ Hz}$		$J_{5z,6\beta} = 11.1 \text{ Hz}$
	$J_{5\alpha, 6\alpha} = 6 \text{ Hz}$		
Η-6α	2.70 dd		_
	$J_{6\alpha, 6\beta} = 15 \text{ Hz}$		
Η-6β	2.67 dd	<u> </u>	4.35 dd
			$J_{6\beta,7\alpha} = 7.8 \text{ Hz}$
Η-7α	_		4.58 br d
			$J_{7\alpha, 14} < 1 \text{ Hz}$
H-14	2.80.8	8.12 s	7.40 <i>br s</i>
			$W_{1/2} = 3 \text{ Hz}$
H-15	3.59 septet	3.53 septet	2.90 br septet
	$J_{15, 16} = J_{15, 17} = 7.5 \text{ Hz}$	$J_{15,16} = J_{15,17} = 7.2 \text{ Hz}$	$J_{15,16} = J_{15,17} = 6.9 \mathrm{Hz}$
			$J_{15,14} < 1 \text{ Hz}$
Me-16†	1.28 d	1.28 d	1.04 d
Me-17†	1.27 d	1.28 d	1.03 d
Me-18‡	0.86 s	1.59 s	1.23 s
Me-19‡	0.91 s	1.63 s	1.37 s
Me-20	1.55 s	1.89 s	1.50 s

^{*}Spectral parameters were obtained by first order approximation.

treatment with ethereal diazomethane, which yielded a compound identical in all respects (mp, mmp, [a]_p, IR, UV, ¹H NMR and mass spectra) with natural cryptojaponol (2) [12–14], a diterpenoid whose structure is firmly established by partial syntheses [9, 19]. Demethyl-cryptojaponol (1) has been previously synthesized, but it was transformed without characterization into cryptojaponol [9, 19].

Another of the new diterpenoids, 14-deoxycoleon U (3), had a molecular formula $C_{20}H_{20}O_4$, and its IR spectrum showed phenolic (3530, 3390, 3240–2500 broad, cm⁻¹) and additionally conjugated aryl-ketone (3030, 1630, 1585, 1555 cm⁻¹) absorptions. The UV spectra of compound 3, obtained in neutral medium and after addition of alkali, aluminium chloride, aluminium chloride plus hydrochloric acid, and boric acid (Table 1), showed that

^{*}Decomposition; addition of hydrochloric acid does not regenerate the spectrum in ethanol.

[†]No variation was observed.

^{†,‡}Assignments bearing the same sign may be reversed.

[§]Overlapped with the H-15 signal.

the molecule of this diterpenoid possessed o-diphenol and diosphenol groupings, because the observed band shifts in different media are characteristic of these chromophores [16]. The ¹H NMR spectrum of 14-deoxycoleon U (Table 2) was almost identical with the spectrum of coleon U (6, 11, 12, 14-tetrahydroxy-5, 8, 11, 13-abietatetraen-7one) [18], the only difference was the presence in the spectrum of compound 3 of a one-proton singlet signal at δ 8.12, which must be assigned to the C-14 aromatic proton [11–15]. Thus, structure 3 was established for this new diterpenoid and this structure is in agreement with its ¹³C NMR spectrum (Table 3). Effectively, the C-1-C-7, C-10, and C-18-C-20 carbon atom resonances of compound 3 were almost identical with those reported for coleon U [18], whereas their C-8, C-9 and C-11-C-17 carbon atom resonances were the same as the ones $(20,6\beta-\text{epoxy-}6\alpha,11,12$ observed carnosolone in trihydroxy-8,11,13-abietatrien-7-one) [20]. Finally, a normal abietane absolute configuration was established for 14-deoxycoleon U (3), because the variation of the optical rotations of this compound and its di- and triacetyl derivatives (4 and 5, respectively, see Experimental) is similar to that observed for coleon U and its acetates, and the normal abietane absolute configuration of these last compounds is well-known [18].

Table 3. 13 C NMR chemical shifts (in δ -values from TMS) of compound 3*

C 1	20.4 .4	C 11	120.0
C-1	30.4 t †	C-11	139.9 s
C-2	18.2 t	C-12	149.7 s
C-3	37.0 t	C-13	135.4 s
C-4	36.6 s	C-14	116.4 d
C-5	143.8 s‡	C-15	27.6 d
C-6	143.9 s‡	C-16	22.8 q§
C-7	180.5 s	C-17	23.0 q§
C-8	121.3 s	C-18	28.1 q
C-9	142.3 s‡	C-19	27.6 q
C-10	41.3 s	C-20	28.4 q

- *At 20.15 MHz, in pyridine-d₅ solution.
- †SFORD multiplicity.
- ‡, §, || Assignments bearing the same sign may be interchanged.

The last diterpenoid was named salviphlomone. It possesses the structure depicted in 6, which is in complete agreement with the following results. Salviphlomone (6) had a molecular formula C₂₀H₂₈O₄, and its IR spectrum showed hydroxyl (3570, 3460 cm⁻¹) and o-benzoquinone $(1677, 1660, 1605, 1580 \,\mathrm{cm}^{-1})$ absorptions [21]. The presence in salviphlomone of an o-benzoquinone moiety was also confirmed by its UV spectrum (Table 1) which was identical to that of 3,5-di-t-butyl-1,2-benzoquinone [21], and by the fact that in its mass spectrum there were two ions at m/z 330 $[M-2]^+$ and 334 $[M+2]^+$, which are only typical of o-quinonoid compounds [21]. The ¹H NMR spectrum of salviphlomone (6) showed signals of an isopropyl group and three methyl groups identical with those found in the other diterpenoids (Table 2). A one-proton doublet of triplets signal at δ 2.76 was assigned to the C-1 β equatorial proton [17, 18] and three signals at $\delta 1.53$ (d), 4.35 (dd) and 4.58 (br d) were attributed to the C-5-C-7 protons, respectively. The coupling values of these protons ($J_{5,6} = 11.1$ Hz, $J_{6,7} = 7.8$ Hz) are only compatible with a 5α -H, 6β -H, 7α -H arrangement in abietanes with an aromatic or pseudoaromatic C ring [22]. Finally, the C-14 quinonoid proton of salviphlomone (6) appeared in its ¹H NMR spectrum at δ 7.40 as a broad singlet ($W_{1/2} = 3$ Hz) and, on irradiation at this field, the lines of the signals at δ 4.58 (H- 7α) and 2.90 (H-15) were clearly sharpened. Thus, this quinonoid proton must be allylic to both C-7 and C-15 protons, the o-quinone group of salviphlomone must be at the C-11 and C-12 positions of an abietane skeleton and, in consequence, only structure 6 is likely for this new diterpenoid.

The absolute configuration of salviphlomone was not ascertained. However, compound 6 is believed to belong to the normal series like the other diterpenoids co-occurring in the same species.

Two of the previously known diterpenoids now isolated from the root of Salvia phlomoides, taxodione and taxodone, possess tumor-inhibitory activity against the Walker 256 intramuscular carcinosarcoma in rats [11], and this is the second report of their occurrence in plants, although the 14-methoxy or hydroxy derivatives of taxodione have been previously isolated from Labiatae species [23, 24]. From a chemotaxonomic point of view it is important to note that only two (royleanone and 7α -acetoxyroyleanone) of the 11 diterpenoids isolated from Salvia phlomoides have been previously found in plants belonging to the Labiatae family.

EXPERIMENTAL

Mps are uncorr. For general details on methods see refs. [1-4]. Assignments of ¹³C NMR chemical shifts were made with the aid of off-resonance and noise-decoupled ¹³C NMR spectra. Plant materials were collected in April 1982, near Zaorejas (Guadalajara, Spain), and voucher specimens were deposited in the Herbarium of the Faculty of Pharmacy (Madrid 'Complutense' University).

Extraction and isolation of the diterpenoids. Dried and finely powdered S. phlomoides Asso. roots (370 g) were extracted with Me₂CO (2.5 l.) at room temp. for 1 week. After filtration, the solvent was evaporated yielding a red gum (17 g) which was subjected to dry CC over Si gel (400 g, Merck No. 7734, deactivated with 10% H₂O). Elution with n-hexane and nhexane-EtOAc mixtures yielded the following compounds in order of elution: 8,13-abietadiene (6 mg) [5, 6], 8,11,13-abietatriene (53 mg) [7-9], royleanone (60 mg) [10, 11], taxodione (1.7 g) [11], 7α-acetoxyroyleanone (210 mg) [10], cryptojaponol (2, 26 mg) [12-14], taxodone (75 mg) [11], sugiol (48 mg) [11, 13-15], demethylcryptojaponol (1, 460 mg), 14-deoxycoleon U (3, 280 mg) and salviphlomone (6, 145 mg). The previously known diterpenoids were identified by their physical (mp, $[\alpha]_D$) and spectroscopic (IR, UV, 1H NMR, MS) data and, in some cases, by comparison with authentic samples.

Demethylcryptojaponol (1). Mp 184–187° (from EtOAc–nhexane, pale yellow plates); $[\alpha]_D^{26} + 31.2^\circ$ (CHCl₃; c0.635); IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 3590, 3380 br, 3020, 3010, 2970, 2940, 2880, 2860, 1655, 1635, 1570, 1465, 1330, 1250, 1205, 1145, 1110, 995, 895, 890, 790, 780; UV: see Table 1; ¹H NMR: see Table 2; EIMS (direct inlet) 75 eV m/z (rel. int.): 316 [M]⁺ (100), 301 (79), 260 (49), 247 (37), 233 (67), 231 (73), 219 (79), 205 (40), 203 (23), 189 (22), 179 (27), 128 (22), 115 (29), 105 (16), 91 (30), 83 (23), 77 (29), 69 (51), 55 (47). (Found: C, 75.52; H, 8.94. $C_{20}H_{28}O_{3}$ requires: C, 75.91; H, 8.92 %.)

Cryptojaponol (2) from demethylcryptojaponol (1). Ethereal

Compound	589 (D) nm	578 nm	546 nm	436 nm	365 nm	c
3	+ 26.8°	+ 25.8°	+ 20.9	120 3°		0.620
4	+ 32.1°	+ 31.5°	+ 28.1	- 92.6°		0.020
5	+ 79.5°	$+83.3^{\circ}$	+93.4°	+ 140.1°	−111.6°	0.516

Table 4. Optical rotations of $[\alpha]^{24}$ compounds 3-5 (solvent CHCl₃)

 CH_2N_2 treatment of 1 (80 mg) for 3 hr at 5° yielded 78 mg of a compound identical (mp, mmp, $[\alpha]_D$, TLC, IR, UV, 1H NMR and MS) in all respects with natural cryptojaponol (2) [9, 12–14, 19].

14-Deoxycoleon U (3). Mp 205–207° (EtOAc-n-hexane, yellow prisms); IR $v_{\rm max}^{\rm KBr}$ cm $^{-1}$: 3530, 3390, 3240–2500 br, 3030, 3000, 2970, 2940, 2880, 1630, 1585, 1555, 1470, 1415, 1380, 1345, 1300, 1275, 1190, 1140, 1065, 1000, 915, 880, 805, 790; UV: see Table 1; 1 H NMR: see Table 2; 13 C NMR: see Table 3; EIMS (direct inlet) 75 eV m/z (rel. int.): 330 [M] $^{+}$ (47), 315 (12), 302 (4), 287 (15), 274 (12), 269 (5), 261 (68), 260 (100), 248 (16), 247 (13), 245 (18), 233 (13), 231 (13), 219 (11), 217 (10), 203 (6), 128 (7), 115 (7), 91 (6), 83 (7), 77 (5), 69 (5), 55 (7). (Found: C, 73.02; H, 8.00. $C_{20}H_{26}O_4$ requires: C, 72.70; H, 7.93 %.)

11,12-Diacetyl-14-deoxycoleon U (4) and 6,11,12-triacetyl-14-deoxycoleon U (5) from compound 3. Ac₂O-pyridine treatment of 3 (135 mg) for 24 hr at room temp. yielded a mixture of compounds 4 and 5, which was subjected to CC (Si gel, n-hexane-EtOAc, 2:1, as eluent) to yield the diacetate 4 (68 mg, less polar component) and the triacetyl derivative 5 (61 mg).

Compound 4. An amorphous solid, mp 80–85°; IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3390, 2980, 2950, 2890, 1780, 1645, 1635, 1610, 1565, 1445, 1375, 1325, 1200, 1180, 1140, 1025, 930, 860, 810, 800, 760; UV $\lambda_{\text{max}}^{\text{EIOH}}$ nm (log ε): 217 (4.24), 265 (4.02), 315 (4.05); ¹H NMR (90 MHz, CDCl₃): δ 8.11 (1H, s, H-14), 7.03 (1H, s, disappears after D₂O exchange, –OH), 2.95 (1H, septet, J=7 Hz, H-15), 2.33 and 2.30 (3H each, s, two –OAc), 1.25 and 1.21 (3H each, d, J=7 Hz, 3H-16 and 3H-17), C–Me singlets at 1.59 (3H), 1.47 (3H) and 1.43 (3H); EIMS (direct inlet) 75 eV m/z (rel. int.): 414 [M] ⁺ (18), 372 (17), 344 (8), 330 (15), 303 (25), 261 (30), 260 (32), 245 (12), 232 (10), 217 (10), 128 (10), 115 (10), 91 (8), 83 (14), 77 (12), 69 (20), 55 (18), 43 (100). (Found: C, 69.14; H, 7.37. $C_{24}H_{30}O_{6}$ requires: C, 69.54; H, 7.30 %.)

Compound 5. Mp 154–156° (Et₂O–n-hexane); IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3040, 3020, 2995, 2950, 2890, 1775, 1670, 1630, 1615, 1565, 1470, 1430, 1395, 1370, 1340, 1265, 1210, 1170, 1110, 1060, 1025, 1000, 925, 895, 865, 815, 797, 760, 696; UV $\lambda_{\text{max}}^{\text{EIOH}}$ nm (log ε): 220 (4.03), 263 (4.10), 279 (4.08); ¹H NMR (90 MHz, CDCl₃): δ 8.09 (1H, s, H-14), 2.93 (1H, septet, J=6.5 Hz, H-15), 2.35 (6H, s, two –OAc), 2.30 (3H, s, one –OAc), 1.23 and 1.20 (3H each, d, J=6.5 Hz, 3H-16 and 3H-17), C–Me singlets at 1.63 (3H) and 1.38 (6H); EIMS (direct inlet) 75 eV m/z (rel. int.): 456 [M] ⁺ (15), 414 (20), 372 (18), 345 (7), 344 (6), 330 (12), 303 (26), 261 (21), 260 (20), 231 (5), 128 (5), 115 (5), 105 (5), 91 (6), 83 (7), 77 (5), 69 (8), 55 (7), 43 (100). (Found: C, 68.66; H, 7.17. C₂₆H₃₂O₇ requires: C, 68.40; H, 7.07 %.)

Optical rotations of compounds 3-5. See Table 4.

Salviphlomone (6). Mp 163–165° (EtOAc–n-hexane, deep red needles); $[\alpha]_{25}^{25} - 330.4$ °, $[\alpha]_{578} + 69.6$ °, $[\alpha]_{546} + 1217.4$ ° (MeOH; c 0.232); IR $v_{\rm max}^{\rm KBr}$ cm $^{-1}$: 3570, 3460, 3020, 2970, 2930, 2870, 1675, 1653, 1580, 1465, 1395, 1258, 1120, 1090, 1020, 1010, 990, 920, 890, 760; $v_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3590, 3400, 1677, 1660, 1605, 1580; UV: see Table 1; 1 H NMR: see Table 2; EIMS (direct inlet) 75 eV m/z (rel. int.): 334 [M + 2] $^+$ (18), 332 [M] $^+$ (37), 330 [M $^-$ 2] $^+$ (8), 317 (14), 314 (8), 303 (68), 299 (13), 286 (50), 271 (39), 247 (50), 231 (37), 219 (42), 215 (34), 203 (32), 179 (39), 161 (26),

145 (26), 129 (24), 128 (29), 121 (24), 115 (39), 109 (32), 105 (34), 91 (66), 77 (63), 69 (81), 55 (100); 10 eV m/z (rel. int.): 332 [M] $^+$ (100), 317 (19), 316 (22), 314 (18), 303 (72), 286 (12), 247 (12), 146 (17), 109 (10). (Found: C, 72.60; H, 8.58. $C_{20}H_{28}O_4$ requires: C, 72.26; H, 8.49 $^\circ$ ₀.)

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